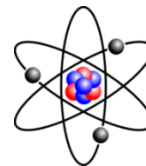


Name:



Physical Science MATH Light Assignment

Directions: Sign the pledge; write the formula that you use for each question; show all your work; circle your final answer.

I promise that I have neither given nor received any information beyond that permitted by the instructor. **Signed:**

$c = \lambda f$
 $E = hf$
c: speed of light 3.00×10^8 m/s
 λ : Wave Length in m
f: Frequency in Hz or s^{-1}
h: Plank's Constant 6.63×10^{-34} joule sec
E: Energy in J/s or kgm^2/s^3

Color	Wave Length (λ) Write in Sci Notation	Frequency in Hz Write in Sci Notation	Energy in Joules Write in Sci Notation
Violet (Example)	400 nm $= 4.00 \times 10^{-7}$	$c = \lambda f$ $3.00 \times 10^8 = 4.00 \times 10^{-7} f$ $3.00 \times 10^8 / 4.00 \times 10^{-7} = f$ $0.75 \times 10^{15} = $ $7.5 \times 10^{14} = f$	$E = hf$ $E = 6.63 \times 10^{-34} (7.5 \times 10^{14})$ $E = 4.97 \times 10^{-19}$
Indigo	445 nm		
Blue	475 nm		
Green		5.88×10^{14} Hz	
Yellow		5.26×10^{14} Hz	
Orange			3.37×10^{-19}
Red			3.26×10^{-19}
Cell Phone	12 cm		
Typical X-Ray	0.01 nm		

